

Wednesday 2/21/12

- Turn in your writing assignment!
- Anything due tomorrow?

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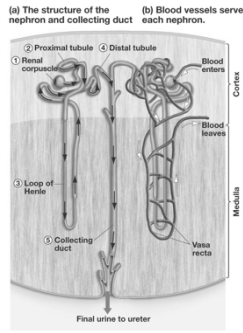
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### Review

1. Renal Corpuscle
  - Filtration
2. Proximal Tubule
  - Reabsorption
3. Loop of Henle
  - Establishing an osmotic gradient



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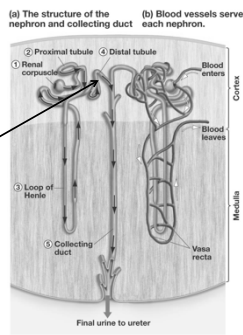
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### Review

- What solutes are left?
  - Urea, other wastes
- Fluid entering the distal tubule is slightly hypotonic to blood.
- Fluid osmolarity is always consistent



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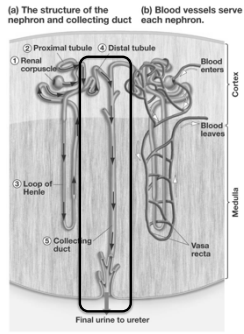
### The Distal Tubule and Collecting Duct

4. Distal tubule

5. Collecting duct

Activity changes in response to osmotic stress

Both respond to hormones




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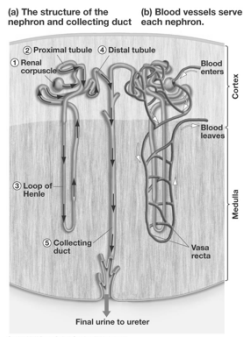
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### The Distal Tubule and Collecting Duct

• Normal functions

• Distal Tubule:  
– Reabsorb  $\text{Na}^+$ ,  $\text{Cl}^-$ ,  $\text{H}_2\text{O}$

• Collecting Duct  
– Normally impermeable




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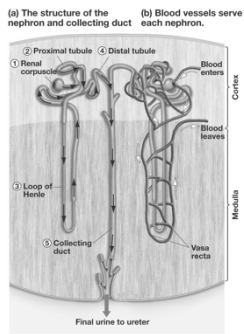
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### The Distal Tubule and Collecting Duct

• Regulated by hormones when...

• Low blood  $\text{Na}^+$   
– Too much water

• Dehydration  
– Too little water




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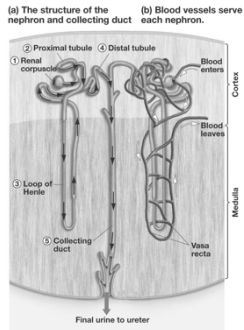
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### The Distal Tubule and Collecting Duct

- Low blood  $\text{Na}^+$ 
  - Too much water
  - **aldosterone**
  - More  $\text{Na}^+$  pumps activated, more  $\text{Na}^+$  reabsorbed




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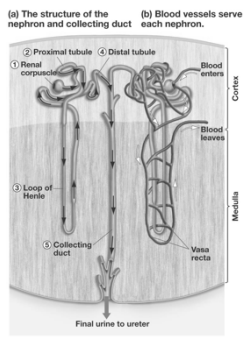
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### The Distal Tubule and Collecting Duct

- Dehydration
  - Too little water
  - **ADH** (antidiuretic hormone)
  - Aquaporins inserted in plasma membrane
  - Membrane more permeable to urea (?)




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### The collecting duct and ADH

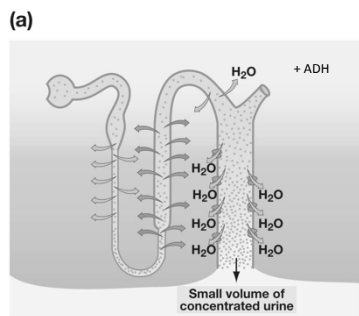


Fig. 42.18 pg. 953

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### Questions

- What is the major difference between the first 3 segments of the nephron and the last 2 segments?
- Which segment of the nephron is inactive under normal conditions?

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### Write this down.

- Describe what will happen in the nephron in response to a person drinking too much alcohol. Keep in mind that alcohol is a diuretic.
- Describe what will happen in the nephron in response to someone who was just given ecstasy and a case of bottled water.



<http://www.alcoholbars.com/wp-content/uploads/2011/03/Why-do-People-Drink-Alcohol-1.jpg>  
<http://www.meaningandhappiness.com/pictures/blog/happy-fool.jpg>

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## Photosynthesis

- Properties of light and pigments
- Photosystem II
- Photosystem I
- Calvin Cycle
- Adaptive variations




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## Light Spectrum

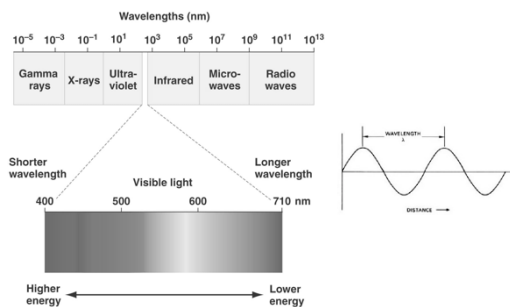


Fig 10.4 pg 202

<http://www.davidarling.info/encyclopedia/W/wavelength.html>

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## Pigments involved in photosynthesis

Pigments	Wavelengths
Chlorophyll a	Absorbs blue and red
Chlorophyll b	Absorbs blue and red
Carotenoids (e.g. $\beta$ -carotene)	Absorbs UV, stabilizes unpaired electrons and free radicals (Sunscreen)
	Absorbs blue and green

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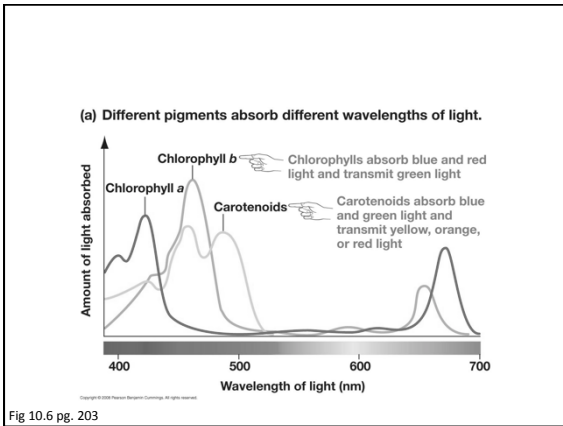
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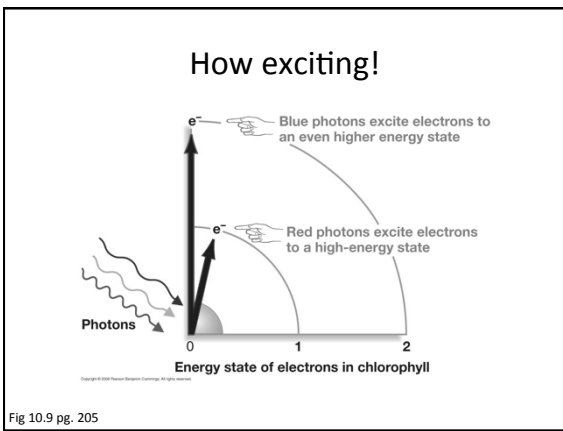
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**Exciting Consequences!**

- When an electron in a chlorophyll molecule is excited, it can cause:
- Fluorescence
- Resonance
- Redox Reaction
- Depends on location of chlorophyll molecule

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### Exciting Consequences!

- Solitary chlorophyll molecule
  - Fluorescence (release of light and heat)
  
- Antenna Complex
  - Resonance (vibration)
    - Transfer of energy (not the electron) until it reaches...
  - Redox reaction
    - Reaction center

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### Exciting Consequences!

The diagram illustrates three possible outcomes for an excited electron in a chlorophyll molecule:

- FLUORESCENCE or RESONANCE:** The electron drops back down to a lower energy level, and heat and fluorescence are emitted. This is labeled "Energy Lost".
- Energy Transfer:** Energy in the electron is transferred to a nearby pigment (the Reaction center). This is labeled "Energy Transferred".
- REDUCTION/OXIDATION:** The electron is transferred to a new compound (the Electron acceptor). This is labeled "Energy Captured".

Fig. 10.11 pg. 206

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### Photosystem II

Pheophytin captures electron, donates it to intermediates (redox) in an electron transport chain

ETC creates a proton gradient

The diagram shows a photon hitting a chlorophyll molecule, which then transfers an electron (e<sup>-</sup>) to pheophytin. The energy level of the electron is shown increasing from a lower state to a higher state.

Fig. 10.13 pg. 208

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## Redox Reactions

- Oxidation - Reduction
- A class of reactions that cause the loss or gain of an electron.
- An electron donor must always be paired with an electron acceptor

http://www.emc.maricopa.edu/faculty/farabee/BIGB/06bookexm.html

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## Proton Motive Force!

(a) In photosystem II, excited electrons feed an electron transport chain.

(b) Plastocyanin carries protons to the inside of thylakoids, creating a proton-motive force.

Pheophytin serves the same role in photosynthesis as NADH and FAD<sub>2</sub> do in respiration:  
Donating electrons to create a proton gradient

Fig. 10.13 pg 208

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